

Course Syllabus

Teacher: Ms. Cynthia F. Holden Email: cholden@madisoncity.k12.al.us Phone: 256-772-2547 ext. 80230

## I. Course Description:

Pre-AP Chemistry is an accelerated course designed to prepare students for Advanced Placement Chemistry; however, taking AP Chemistry is not mandatory. Chemistry is the science of matter & how it behaves. We will explore the laws of chemistry which govern the behavior of matter & energy. Students will gain knowledge about measurements, chemical formula writing, quantitative theory, chemical reactions, stoichiometry, the periodic table and chemical bonding. Problem solving and technical writing are important facets of this course. Labs and activities will reinforce material covered in lecture. The intent of this course is to prepare students for college level chemistry and continued study in science.

### II. Course Requirements

- The student must have prior teacher approval.
- Students are recommended to have taken or be currently enrolled in Algebra II.
- Students are expected to <u>actively participate</u> in all aspects of the class including, but not limited to discussions, laboratory, reports, projects, homework/class assignments.
- The student is responsible for all text materials & lecture notes.

# III. Course Objectives:

#### Students will:

- Differentiate among pure substances, mixtures, elements, and compounds.
- Describe the structure of carbon chains, branched chains, and rings.
- Use the periodic table to identify periodic trends, including atomic radii, ionization energy, electronegativity, and energy levels.
- Describe solubility in terms of energy changes associated with the solution process.
- Use the kinetic theory to explain states of matter, phase changes, solubility, and chemical reactions.
- Solve stoichiometric problems involving relationships among number of particles, moles, and masses of reactants and products in a chemical reaction.
- Explain the behavior of ideal gases in terms of pressure, volume, temperature, and number of particles.
- Distinguish among endothermic and exothermic physical and chemical changes.
- Distinguish between chemical and nuclear reactions.

# IV. Classroom Expectations:

- 1. Be in your seat and ready to begin when the bell rings.
- 2. Bring all needed materials to class (books, notebook, paper, pencil, etc.) We will work <u>EVERY</u> day. Therefore, it is imperative that each student bring all needed supplies each day.
- 3. Speak at appropriate times using appropriate language & voices. It is extremely important that we respect those who have the floor. If you have a comment or question, please raise your hand to be recognized.
- 4. Be respectful of others & their property!
- 5. Do not bring food or drink into the classroom or lab.
- 6. Go to the restroom between classes & other allotted breaks.
- 7. Dismissal will occur at the end of the period by the instructor.
- 8. Do not leave your seats for any reason without permission from the instructor.
- Obev all school rules.

#### Consequences for Violations of Classroom/School Rules:

- First Offense: Verbal Warning
- Second Offense: After School Detention/Parent or guardian called
- Third Offense: Discipline Referral

<sup>\*</sup>Severe disruptions or offensive behavior will not be tolerated! Students will be sent immediately to a school administrator.



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## V. Student Expectations:

- Students must keep a list of their own grades.
- Students will receive earned grade. <u>Student or parents should not ask for</u> additional points.
- There will be no extra credit assignments i.e. reports, etc. There may be bonus points on tests or daily activities.
- Students are responsible for bringing their materials to class daily (text, notebook, calculator, pen/pencil)
- Students that receive failing grades during the course are urged to consult the teacher.

# VI. BJ Grading Policy:

Grades will be calculated according to BJHS policy:

90%-100+%=A 80%-89%=B 70%-79%=C 65%-69%=D 64% and below= F

Final Exam (Comprehensive)......20% of entire grade

\*Honors courses receive Honors weighted credit per the current grading scale.

## VII. BJ Make-up Test Policy:

- Tests due to excused absences must be made up within 2 weeks. After the time has elapsed, you will receive a zero. Make-up texts will not be given for unexcused absences. Make-up tests are administered on Thursday mornings at 7:15 in the designated science room-N204. Students are responsible for obtaining a pass from the teacher the day before the morning make up date.
- If you are absent for a school sponsored activity, you are responsible for turning in your assignment upon your return.
- It is the student's responsibility to make up all missed tests and all assignments.
- There will be "While you were out Notebook" for missed assignments in the student resource section of the room. Check this notebook for missed assignments & lesson plans.
- Find a student partner to call when you are absent.
- Additionally, students will keep a log of all assignments in their notebooks.

## VIII. Text and Other Required Reading:

Modern Chemistry, Holt Rinehart, & Winston, 2006

# IX. Materials and Supplies Needed:

- 1-1 1/2" binder with dividers. This notebook will be used for this class only.
- Lose leaf paper.
- Pens/Pencils (blue/black ink only)
- #2 pencils w/eraser. You may take notes in pen, but all test & problems must be done in pencil.
- Scientific calculator is required! Each student must bring their personal calculator to class daily.



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# Materials and Supplies Needed:

- Textbook
- Pre-AP Chemistry~Laboratory Manual must be purchased for a fee of \$10.
- 1 roll of paper towels.
- 1 ream of paper (white or colored)

Special Note to Parents: The Chemistry Department is requesting a \$20 donation towards needed supplies. Although this donation is optional, it would be greatly appreciated and would allow additional enhancement of your child's education.

### X.

### Homework/ Classwork:

- Assignments will be taken up & checked daily. Assignments will be checked for neatness, completeness, & correctness.
- It is imperative to complete homework and daily assignments. Many of the concepts taught are fundamental to understanding more complex ideas in subsequent lessons.
- All work must be done in "your own handwriting". Xeroxed work will not be accepted.
- Notebook assignments must be given a title at the top of the page, page #, & problem set. (Example: Chapter 1 Review Questions pg.14 #s 1-15)
- Missed assignments due to absence must be completed within 2 days. The student is responsible for completing and turning in the assignment to the teacher.

#### XI.

#### **Tardiness:**

- Students should be seated and working on bell activity when the bell rings.
- If the bell rings when you are entering the room, you will sign a tardiness form and receive detention the following day.
- If you are detained in another class by a teacher, simply bring a note from that teacher, to prevent excuse the tardy.

#### X.

## Rules for Personal Laptop Usage in the Classroom:

- 1. Under no circumstances are student laptops to be wired to the network.
- 2. Students may not print from their laptops.
- 3. Laptops in the classrooms will be used for taking notes, research, word processing, and creation of materials for projects
- 4. Neither the teacher, nor the school is responsible for broken, stolen, or lost laptops
- 5. If you are asked by the teacher to put away your laptop, please do so- just as you may be asked to put away something else that is distracting or inappropriate.
- 6. Laptops will remain in the student's bag until instructor grants permission for student to open and operate it.
- 7. Headphones are not to be used with laptops during class unless approved by the instructor.

## Additionally,

- The academic miscontent policy of the school will be followed in this course.
- The attendance policy of the school will be followed for this course.
- All requests for accommodations, for this course or any school event, are welcomed from students and parents.



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18 – WEEK PLAN						
Week 1	Intro to Chemistry (Lab Safety) & Classification/Investigation of Matter					
Week 2	Measurements and Calculations					
Week 3	Atomic Structure/Periodic Table/Electron Configurations/ Nuclear RXNs					
Week 4	Chemical Bonding/Formulas & Compounds					
Week 5	Chemical Bonding/Formulas & Compounds					
Week 6	Chemical Reactions and Equations					
Week 7	Chemical Reactions and Equations					
Week 8	Chemical Measurements (Mole Concept)					
Week 9	Stoichiometry					
Week 10	Stoichiometry					
Week 11	Analytical Chemical Techniques					
Week 12	Solids, Liquids & Organic Chemistry					
Week 13	Gases					
Week 14	Solutions					
Week 15	Solutions					
Week 16	Acids & Bases					
Week	Thermodynamics/Rates of RXNs					
Week 18	Equillibria  Review & Final Examination (Comprehensive!!!)					

<sup>\*</sup> The course outline is tentative. Dates are subject to change.



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**********	********Detac	h & Return******* Signed	**********	·**********	**
Student's Name	Grade				
Parent's Name					
Daytime Number	Cell Number:				
Parent's Email Address					
** Email Address Parents, please provide an accurate updates may be distributed.	email address t	hat you check regula	rly. A distribution list will be set u	лр, so that e	email
***Google Classroom/OWL Online We Google Classroom is an on-line tech student. Due to the lack of funding, student. To prepare for this, please indicate y	nology tool use many of the as	signments/handouts			
**** Remind 101 Remind 101 is a website that pro	vides a safe wa	ay for teachers to te	xt message or email students and	l parents.	
Internet Access:	_Yes	No			
Print Capability:	Yes	No			
Texting Capability:	Yes	No			
***** Laboratory Information:					
Allergic Reactions:	Yes	No			
If so, please list specific Allergi	es/:				
List any medical conditions of	which teacher s	hould be aware:			_
Do you wear contact lenses? _	Yes	No			
Are you color blind?	Yes	No			
I Student's Name (Print Please)	, acknowledge	e that I have received	/printed a copy of Ms. Holden's c	ourse syllab	ous &
Laboratory contract for Chemistry I (_ agree to abide by them.	<del>-</del>	academic school	year). I understand these rules ar	ıd expectati	ions and
Parent's Signature	Date	Student's Signat	ure	Date	Page